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Liquid Crystals

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Preliminary Communications First liquid crystalline derivatives of umbelliferyl β -D-glucoside [1]

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Preliminary Communications

First liquid crystalline derivatives of umbelliferyl β -D-glucoside [1]

by DIRK BLUNK, KLAUS PRAEFCKE* and GÜNTER LEGLER†

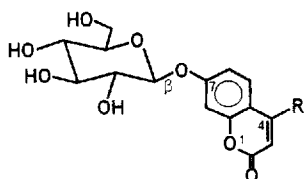
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Three examples of a new series of carbohydrate liquid crystals incorporating a coumarin unit are reported. A steric effect influencing their smectic A phase formation is discussed in comparison with structurally unaffected amphiphilic carbohydrate-based mesogens.

In the course of our work on natural product liquid crystals [2–5], mainly on multihydroxy amphiphiles, we had access to 4-alkyl derivatives of umbelliferyl β -D-glucoside (**1 a–c**, 4-alkyl-7-(β -D-glucopyranosyloxy)-2H-1-benzopyran-2-ones), and were tempted to examine the anticipated thermomesogenic properties of these bioactive representatives.



1 a: R = C₇H₁₅
1 b: R = C₉H₁₉
1 c: R = C₁₃H₂₇

The molecular structure of the 4-alkyl derivatives **1 a–c** of umbelliferyl β -D-glucoside, investigated here.

The middle part of this type of glucoside **1**, the rigid and relatively flat coumarin unit, is flanked by a hydrophilic tetrol (carbohydrate) head group and a lipophilic tail *R* causing a somewhat bent molecular shape. Despite the humpbacked situation created by positions 1 to 3 of the coumarin sub-structure, our compounds can apparently exist in a conformation that is stretched enough to allow the systems to exhibit mesogenity on heating. Naturally, compounds without such a steric burden, for example, the closely related 1-*O*-trans-4'-alkylcyclohexyl or 1-*O*-4'-substituted-phenyl derivatives of β -D-glucopyranoside, are much better favoured to be thermotropically liquid crystalline over considerably wider ranges of temperature [6]. For instance, whereas **1 a** is not thermomesogenic, the

4-octyl- and 4-heptyloxy-phenyl β -D-glucosides [6] are; interestingly, the melting points of these three carbohydrate aryl ethers are nearly identical (107–108°C). Our recent findings on the liquid crystalline behaviour of the three derivatives **1 a–c** of umbelliferyl glucopyranoside, as the parent compound of a new mesogen family, are reported here.

The synthesis of **1 a–c** has been described elsewhere [7], but relevant information is given later in this article; their structures were proved by IR and NMR spectroscopy.

Whereas the heptyl derivative **1 a** is not liquid crystalline, neither thermo- nor lyo-tropically (in water), both the nonyl and tridecyl homologues **1 b** and **1 c** are—see the table. Miscibility studies (contact method) of the mesophases of **1 b** and **1 c** with 4-*O*-dodecyl-*myo*-inositol

The phase transition data* for the three 4-alkyl derivatives (**1 a–c**) of umbelliferyl β -D-glucoside; the type of thermotropic mesophase exhibited is smectic A(S_A).

Tetrol	Alkyl, <i>R</i>	m.p. (ΔH)	cl.p. (ΔH)
1 a	Heptyl	107.1/107.2 (64.3)	Not liquid crystalline
1 b	Nonyl	159.5/163.2 (33.5)	{158.8/158.8 (2.2)}
1 c	Tridecyl	163.0/163.4 (32.9)	200.0/198.0 (1.3)

* The temperatures for the melting and clearing processes (m.p. or cl.p., respectively) are given in °C and were those obtained by polarizing microscopy/differential scanning calorimetry (ΔH = transition enthalpy is given in kJ mol⁻¹); the data marked in { } relate to a monotropic transition. Melting points of **1 a** and **1 b** determined earlier using a normal m.p. apparatus used in chemical laboratories can be looked up in reference [7]. The new compound **1 c** is the first enantiotropic member of this novel series of mesogens.

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[2(d)] or 1-*O*-octyl- β -D-glucopyranose [8]—two typical members of the so-called single tailed family of carbohydrate liquid crystals—show clearly that they both are of the smectic A type. Between lipophilized glass plates, samples of our umbelliferyl glucopyranoside derivatives **1b** and **c** form focal-conic textures, including wide homeotropic areas, on cooling from their isotropic phases. In contact preparations with water, myelin structures are observable, indicating the formation of a lyotropic mesophase, probably lamellar in type.

Microscopic investigations were carried out with a Leitz Laborlux 12 Pol polarizing microscope equipped with a Mettler FP82 hot stage. The DSC measurements were performed with a Mettler TA 3000/DSC 30 S with GraphWare TA 72. The heating/cooling rates applied in both methods were 5 K min⁻¹.

The lipophilized microscopic slides and cover glasses were prepared by treating the carefully cleaned glass plates in a sealed vessel with some drops of dichlorodimethylsilane for 2 hours.

4-Tridecylumbelliferyl β -D-glucopyranoside [9] (**1c**). Ethyl 3-oxohexadecanoate [10] (10 g, m.p. 41–42°C) was condensed [11] with resorcinol (4.5 g) to give 4-tridecylumbelliferone (5.5 g, m.p. 98–100°C) after recrystallization from ethyl acetate/ether/ethanol 1/1/1. Its reaction with 1-*O*-(2,3,4,6-tetra-*O*-acetyl- α -D-glucopyranosyl) trichloroacetamide in dichloromethane, using BF₃Et₂O as catalyst [12], gave, after purification by MPLC on silica gel in toluene/ethyl acetate 95/5 → 60/40, 4-tridecylumbelliferyl β -D-glucopyranosyl tetra-acetate in 35 per cent yield. The free glucoside was obtained by catalytic deacetylation (sodium methanoate in methanol) and recrystallization from ethanol/water. IR (Beckman I.R. 9, KBr, cm⁻¹): ν = 3405 (OH, very broad), 1696 (C=O), 1612 (C=C), 1073 (aliph. C–O). ¹H NMR (Bruker AM 400, CDCl₃/CD₃OD, ppm): δ = 0.68 (t, 3 H, J = 7 Hz; CH₃), 1.03–1.28 (m, 20 H; 10 alkyl-CH₂), 1.50 (tt, 2 H, each J = 7 Hz; β -CH₂), 2.57 (t, 2 H, J = 7 Hz; α -CH₂), 3.28–3.40, 3.58–3.63 and 3.69–3.74 (3m, 4 H, 1 H and 1 H, respectively, glucosyl-Hs without assignment), 4.81 (d, 1 H, J = 8 Hz; anomeric H), 5.98 (s, 1 H; 3-H), 6.84–6.88 and 7.40–7.44 (2m, 2 H and 1 H, respectively;

5-H, 6-H and 8-H without assignment). Our phase transition data for **1c** are compiled in the table.

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